



Bihar Public Service Commission, Patna

Advt. No. 33/2024, Mechanical Engineering - I, Paper - V

**Assistant Engineer, Mechanical Written (Objective)
Competitive Examination**

Final Answer key (Set- A)

Question No.	Ans. Key	Question No.	Ans. Key
1	C	26	C
2	C	27	C
3	D	28	C
4	D	29	C
5	B	30	C
6	A	31	B
7	D	32	D
8	D	33	A
9	C	34	A
10	B	35	Deleted
11	D	36	C
12	B	37	B
13	B	38	A
14	D	39	D
15	C	40	B
16	D	41	B
17	D	42	B
18	C	43	D
19	C	44	D
20	A	45	A

Question No.	Ans. Key
21	A
22	B
23	B
24	C
25	C

Question No.	Ans. Key
46	D
47	B
48	B
49	B
50	A

(Set - B)

Question No.	Ans. Key	Question No.	Ans. Key
1	B	26	C
2	C	27	B
3	B	28	C
4	B	29	A
5	A	30	A
6	B	31	A
7	D	32	B
8	A	33	D
9	A	34	D
10	D	35	C
11	C	36	B
12	C	37	C
13	Deleted	38	C
14	A	39	D
15	D	40	C
16	D	41	D
17	C	42	A
18	B	43	B
19	B	44	C
20	D	45	A
21	C	46	A
22	A	47	B
23	A	48	C
24	C	49	B
25	C	50	C

(Set - C)

Question No.	Ans. Key	Question No.	Ans. Key
1	D	26	B
2	C	27	D
3	D	28	C
4	A	29	B
5	A	30	D
6	C	31	B
7	A	32	A
8	C	33	D
9	A	34	C
10	C	35	A
11	B	36	C
12	D	37	D
13	B	38	A
14	D	39	C
15	B	40	C
16	C	41	A
17	C	42	C
18	B	43	C
19	A	44	B
20	A	45	A
21	B	46	D
22	B	47	A
23	B	48	C
24	B	49	Deleted
25	A	50	A

(Set - D)

Question No.	Ans. Key	Question No.	Ans. Key
1	B	26	A
2	C	27	A
3	C	28	D
4	B	29	B
5	A	30	D
6	B	31	A
7	B	32	A
8	D	33	D
9	D	34	D
10	C	35	A
11	B	36	C
12	B	37	C
13	B	38	D
14	D	39	A
15	A	40	C
16	A	41	B
17	C	42	B
18	D	43	C
19	C	44	A
20	D	45	A
21	C	46	D
22	A	47	C
23	B	48	D
24	D	49	D
25	B	50	Deleted

Question sequence as per Set - A				
Set-A	Set-B		Set-C	Set-D
15	19		2	22
35	13		49	50
36	18		48	32
44	23		29	6

Reasons as per Set-A

- Q. No. 15- No Change.
Reason :- $\text{CH}_4 + 2\text{O}_2 \Rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
16 Kg 64 Kg
2Kg $(64/16) \times 2 = 8\text{Kg}$
For 2 Kg CH_4 requires 8 kg of O_2 for complete combustion.
- Q. No. 35- Deleted
Reason :- Option (B) and (D) both are correct. From isenthalpic curve and inversion curve, Option (B) is correct. For liquification of gases under Joule-Kelvin expansion through throttle valve, Option 'D' is correct. Ref. Thermal engineering by P.K.Nag, Six edition, page no. 437 & 647.
- Q. No. 36- No Change.
Reason :- (COP ref.) rev. Carnot = $T_2 / (T_1 - T_2)$ COP decreases with increase in temperature difference (T1- T2), i.e. higher temperature has to be increased and lower temperature has to be decreased.
Ref. Engineering thermodynamics by P.K.Nag, six edition, Page no. 629
- Q. No. 44- No Change.
Reason :- Indicated Thermal efficiency (.3) = $P_m \text{ LAN} / (2 \text{ mf.CV})$
Volumetric efficiency = $V_a / V_s = 0.9$, $V_a = 0.9 V_s$
mf = $0.05 \times 0.9 \times V_s = 0.045 V_s$
Indicated Thermal efficiency (.3) = $P_m \times V_s / (2 \times 0.045 V_s \times 45 \times 10^6)$
 $P_m = 12.15 \text{ bar}$
Ref. I.C. Engines by V.Ganesan, fourth edition, Page no. 25.